

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
26 September 2002 (26.09.2002)

PCT

(10) International Publication Number
WO 02/075835 A2

- (51) International Patent Classification⁷: **H01M 8/10**, B01D 71/06, C08G 65/48, C08J 5/22, H01B 1/12 (74) Agents: NEILL, Alastair, William et al.; Appleyard Lees, 15 Clare Road, Halifax HX1 2HY (GB).
- (21) International Application Number: PCT/GB02/01379 (81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.
- (22) International Filing Date: 21 March 2002 (21.03.2002)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
0107075.4 21 March 2001 (21.03.2001) GB
0123085.3 26 September 2001 (26.09.2001) GB
- (71) Applicant (*for all designated States except US*): VICTREX MANUFACTURING LIMITED [GB/GB]; Victrex Technology Centre, Hillhouse International, P.O. Box 4, Thornton Cleveleys, Lancashire FY5 4QD (GB).
- (72) Inventors; and
- (75) Inventors/Applicants (*for US only*): ANDREWS, Mark, James [GB/GB]; 9 Thorntrees Avenue, Lea, Preston PR2 1PJ (GB). LOCKLEY, John, Edward [GB/GB]; 10 Ross-moyne Road, Lancaster LA1 4SN (GB). WILSON, Brian [GB/GB]; 1 White Lea, Cabus, Garstang, Lancashire PR3 1JG (GB).
- (84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

— without international search report and to be republished upon receipt of that report

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

WO 02/075835 A2

(54) Title: FUEL CELL

(57) Abstract: A fuel cell powered by direct fuel, for example a direct methanol fuel cell, includes a polymer electrolyte membrane which includes a semi-crystalline polymer. Preferred semi-crystalline polymers include first repeat units comprising sulphonated aromatic group containing moieties linked by -SO₂- and/or -CO- and/or -Q-groups, where Q is O or S and second repeat units which include aromatic group containing moieties linked by -CO- and/or Q groups.

FUEL CELL

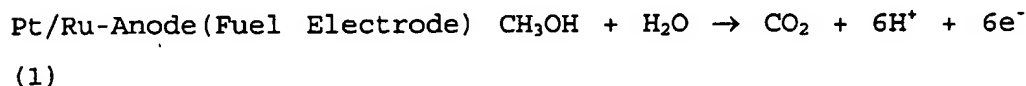
This invention relates to a fuel cell and
5 particularly, although not exclusively, relates to a fuel
cell powered by direct fuel (hereinafter a "FCDF"). The
invention extends to a FCDF, a stack of more than one
FCDF, suitably in series, a method of manufacturing a FCDF
with reduced fuel cross-over, the use of a polymer
10 electrolyte which has at least some crystallinity or is
crystallisable for reducing fuel cross-over in FCDFs,
novel polymer electrolytes per se and methods of preparing
such polymer electrolytes.

15 FCDFs, for example Direct Methanol Fuel Cells
(hereinafter "DMFCs") are known, for example from RM
Moore, Proc. Intersoc, Energy Convers. Eng. Conf. 35th,
Vol.2 1306-1315(2000) (Published Society of Automotive
Engineers), and from US 5 795 496 where they are referred
20 to as "methanol oxidising reducing fuel cells". The term
FCDF is used herein to refer to a fuel cell wherein a
carbon-containing fuel (e.g. methanol) is directly
oxidized at an anode, in contrast to indirect fuel cells
wherein fuel must be "reformed" and "cleaned" to provide
25 hydrogen rich gas "reformat" for use in the fuel cell
system. The term "Direct Methanol Fuel Cell" or DMFC is
used to refer to a fuel cell wherein methanol is directly
oxidized at an anode.

30 Figure 1 of the accompanying schematic drawings
illustrates a DMFC, by way of example.

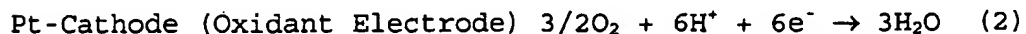
One type of Direct Methanol Fuel Cell (DMFC), shown schematically in Figure 1 of the accompanying diagrammatic drawings, may comprise a thin sheet 2 of a hydrogen-ion conducting Polymer Electrolyte Membrane (PEM) sandwiched on one side by an anode comprising typically a platinum/ruthenium alloy catalyst 3 and a porous carbon electrode 5 and on the other side by the cathode comprising typically a platinum catalyst 4 and a porous carbon electrode 6. The layers 2, 3, 4, 5, 6 make up a Membrane Electrode Assembly (MEA).

In a DMFC, a fuel (methanol) and water mixture is circulated past the anode and the following electrochemical reaction occurs, releasing electrons and protons:

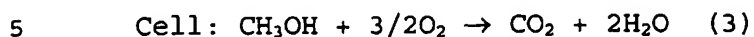


Carbon dioxide produced by the above reaction is withdrawn along with the unreacted fuel and water. The carbon dioxide is separated from the solution in a gas-liquid separator and the fuel and water is recirculated into the cell.

The hydrogen ions or protons migrate through the conducting PEM to the cathode while the electrons are conducted through the external load to the cathode. Simultaneously, an oxidant is introduced at the cathode (oxidant electrode) where the following electrochemical reaction takes place:



The individual electrode reactions described above result in an overall reaction for the exemplary methanol fuel cell given by:



Thus, electrons and protons are consumed to produce water and heat. Connecting the two electrodes through an external circuit causes an electrical current to flow in
10 the circuit and withdraw electrical power from the cell.

In addition to undergoing electro-oxidation at the anode, the liquid fuel which is dissolved in water permeates through solid polymer electrolyte membrane 2 and
15 combines with oxygen on the surface of the cathode electrocatalyst. This process is described by equation 3 for methanol. This phenomenon is termed "fuel crossover". Fuel crossover lowers the operating potential of the oxygen electrode and results in consumption of fuel
20 without producing useful electrical energy. In general, fuel crossover is a parasitic reaction which lowers efficiency, reduces performance and generates heat in the fuel cell. It is therefore desirable to minimise the rate of fuel crossover.

25

A known polymer electrolyte membrane material is a perfluorinated proton-exchange material formed of a copolymer of tetrafluoroethylene and perfluorovinylether sulfonic acid, available from Du Pont under the brand name
30 NafionTM. Nafion 115 has been used extensively for proton-conducting membranes.

Nafion, however, raises its own host of problems. For example, it is very expensive and its methanol permeability allows a substantial amount of fuel crossover across the membrane as described above.

5

It is an object of the present invention to address problems associated with materials used in FCDFs, for example DMFCs.

10 This invention is based on the discovery that the use of semi-crystalline polymers as polymer electrolyte membranes in FCDFs, for example DMFCs, can significantly reduce fuel cross-over in the cell and, therefore, can improve efficiency.

15

According to a first aspect of the invention, there is provided a FCDF, for example a DMFC, said FCDF having a polymer electrolyte membrane (PEM) which includes a semi-crystalline polymer.

20

Surprisingly it has been found that a semi-crystalline polymer, when used as a PEM, allows substantially less fuel cross-over compared to an amorphous polymer of similar type and conductivity. Furthermore, a semi-crystalline polymer has been found to exhibit
25 substantially less fuel cross-over compared to Nafion, a current commercially used material.

The existence and/or extent of crystallinity in a
30 polymer is preferably measured by wide angle X-ray diffraction (also referred to as Wide Angle X-ray Scattering or WAXS), for example as described by Blundell and Osborn (Polymer 24, 953, 1983). Details are provided

in Examples 19 and 20 hereinafter. Alternatively, crystallinity may be assessed by Differential Scanning Calorimetry (DSC).

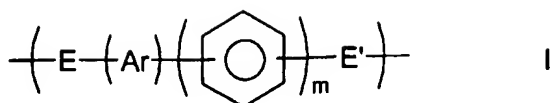
5 The level of crystallinity in said semi-crystalline polymer, suitably prepared as described above, may be at least 0.5%, is preferably at least 1%, is more preferably at least 3% and, especially is at least 5%. In some cases, the crystallinity may be greater than 10% or even
10 greater than 12%. The crystallinity may be less than 20% or less than 15%.

Said semi-crystalline polymer preferably includes a repeat unit which suitably includes aromatic (e.g. phenyl)
15 group containing moieties linked by -CO- and/or Q groups, where Q represents -O- or -S- (especially -O-).

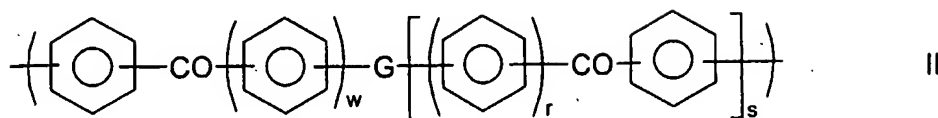
Said semi-crystalline polymer may be made up of a number of repeat units, some of which may be
20 crystallisable or have some crystallinity and some of which will be amorphous. For example, repeat units provided with ion-exchange sites, for example sulphonate groups, will tend to be amorphous, as will repeat units which include bulky groups or -SO₂-. Repeat units which
25 are crystalline or crystallisable suitably include moieties which can be exchanged with ether units in a polyetherketone crystal lattice. Ketone units and/or -S- units may be exchanged and may, therefore, be components of crystalline or crystallisable units.

30

A preferred semi-crystalline polymer is one having a moiety of formula

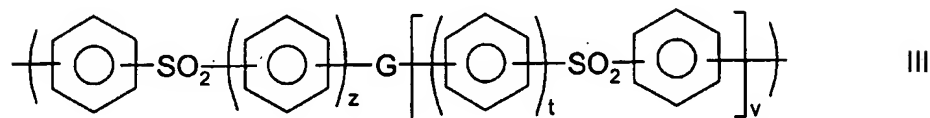


and/or a moiety of formula



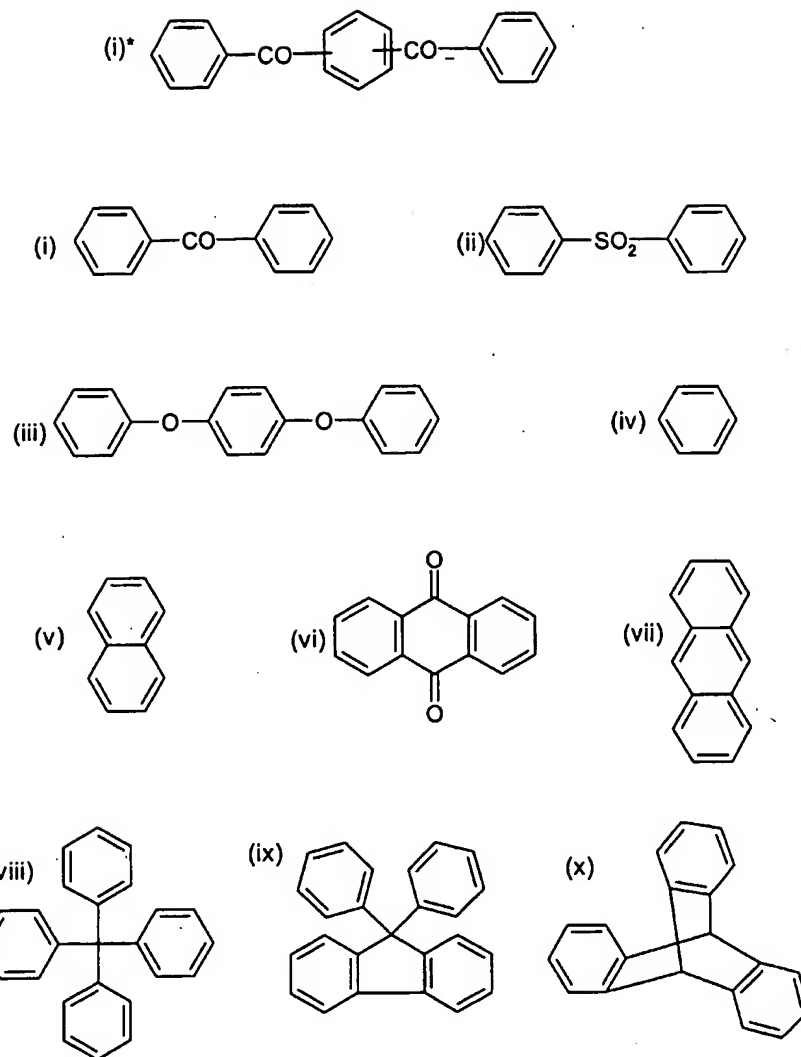
5

and/or a moiety of formula



- 10 wherein at least some of the units I, II and/or III are
 funtionalized to provide ion-exchange sites; wherein the
 phenyl moieties in units I, II, and III are independently
 optionally substituted and optionally cross-linked; wherein
 said material is crystallisable and/or semi-crystalline;
 15 and wherein m,r,s,t,v,w and z independently represent zero
 or a positive integer, E and E' independently represent an
 oxygen or a sulphur atom or a direct link, G represents an
 oxygen or sulphur atom, a direct link or a -O-Ph-O- moiety
 where Ph represents a phenyl group and Ar is selected from
 20 one of the following moieties (i)* or (i) to (x) which is

bonded via one or more of its phenyl moieties to adjacent moieties



5 In (i)*, the middle phenyl may be 1,4- or 1,3-substituted.

Suitably, to provide said ion exchange sites, said
 semi-crystalline polymer is sulphonated, phosphorylated,
 10 carboxylated, quaternary-aminoalkylated or
 chloromethylated, and optionally further modified to yield

-CH₂PO₃H₂, -CH₂NR₃²⁰⁺ where R²⁰ is an alkyl, or -CH₂NAr₃^{x+} where Ar^x is an aromatic (arene), to provide a cation or anion exchange membrane. Further still, the aromatic moiety may contain a hydroxyl group which can be readily elaborated by existing methods to generate -OSO₃H and -OPO₃H₂ cationic exchange sites on the polymer. Ion exchange sites of the type stated may be provided as described in WO95/08581.

Preferably, said semi-crystalline polymer is sulphonated. Preferably, the only ion-exchange sites of said first material are sites which are sulphonated.

References to sulphonation include a reference to substitution with a group -SO₃M wherein M stands for one or more elements selected with due consideration to ionic valencies from the following group: H, NR₄^{y+}, in which R^y stands for H, C₁-C₄ alkyl, or an alkali or alkaline earth metal or a metal of sub-group 8, preferably H, NR₄⁺, Na, K, Ca, Mg, Fe, and Pt. Preferably M represents H. Sulphonation of the type stated may be provided as described in WO96/29360.

Unless otherwise stated in this specification, a phenyl moiety may have 1,4- or 1,3-, especially 1,4-, linkages to moieties to which it is bonded.

Said semi-crystalline polymer may include more than one different type of repeat unit of formula I; more than one different type of repeat unit of formula II; and more than one different type of repeat unit of formula III.

Said moieties I, II and III are suitably repeat units. In the polymer, units I, II and/or III are suitably bonded

to one another - that is, with no other atoms or groups being bonded between units I, II, and III.

Where the phenyl moieties in units I, II or III are
5 optionally substituted, they may be optionally substituted by one or more halogen, especially fluorine and chlorine, atoms or alkyl, cycloalkyl or phenyl groups. Preferred alkyl groups are C₁₋₁₀, especially C₁₋₄, alkyl groups. Preferred cycloalkyl groups include cyclohexyl and
10 multicyclic groups, for example adamantyl. In some cases, the optional substituents may be used in the cross-linking of the polymer. For example, hydrocarbon optional substituents may be functionalised, for example sulphonated, to allow a cross-linking reaction to take
15 place. Preferably, said phenyl moieties are unsubstituted.

Another group of optional substituents of the phenyl moieties in units I, II or III include alkyls, halogens, C_yF_{2y+1} where y is an integer greater than zero, O-R^q (where
20 R^q is selected from the group consisting of alkyls, perfluoralkyls and aryls), CF=CF₂, CN, NO₂ and OH. Trifluormethylated phenyl moieties may be preferred in some circumstances.

25 Said semi-crystalline polymer is preferably not cross-linked. More preferably, ion-exchange sites, e.g. sulphonate groups, throughout the polymer are preferably not cross-linked. Thus, -SO₃H ion-exchange sites on said polymer are preferably free throughout.

30

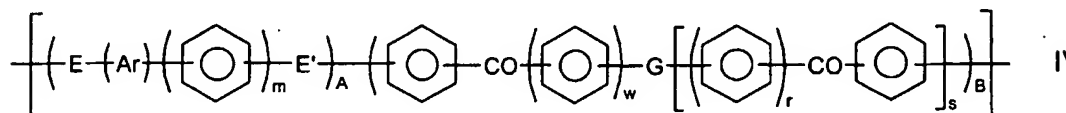
Where w and/or z is/are greater than zero, the respective phenylene moieties may independently have 1,4- or 1,3-linkages to the other moieties in the repeat units

of formulae II and/or III. Preferably, said phenylene moieties have 1,4- linkages.

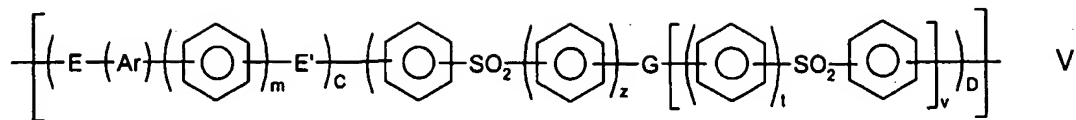
Preferably, the polymeric chain of the semi-crystalline
5 polymer does not include a -S- moiety. Preferably, G represents a direct link.

Suitably, "a" represents the mole % of units of formula I in said polymer, suitably wherein each unit I is the
10 same; "b" represents the mole % of units of formula II in said polymer, suitably wherein each unit II is the same; and "c" represents the mole % of units of formula III in said polymer, suitably wherein each unit III is the same. Preferably, a is in the range 45-100, more preferably in
15 the range 45-55, especially in the range 48-52. Preferably, the sum of b and c is in the range 0-55, more preferably in the range 45-55, especially in the range 48-52. Preferably, the ratio of a to the sum of b and c is in the range 0.9 to 1.1 and, more preferably, is about 1.
20 Suitably, the sum of a, b and c is at least 90, preferably at least 95, more preferably at least 99, especially about 100. Preferably, said polymer consists essentially of moieties I, II and/or III.

25 Said polymer may be a homopolymer having a repeat unit of general formula



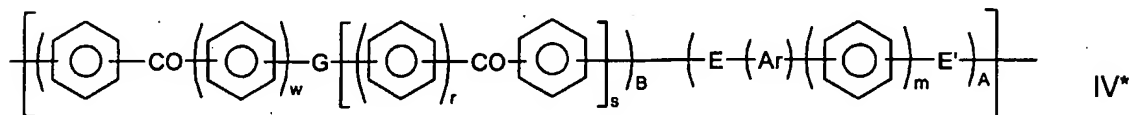
or a homopolymer having a repeat unit of general formula



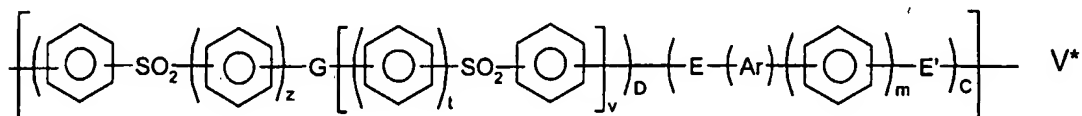
or a random or block copolymer of at least two different units of IV and/or V

- 5 wherein A, B, C and D independently represent 0 or 1 and E, E', G, Ar, m, r, s, t, v, w and z are as described in any statement herein.

As an alternative to a polymer comprising units IV
10 and/or V discussed above, said semi-crystalline polymer may be a homopolymer having a repeat unit of general formula



or a homopolymer having a repeat unit of general formula



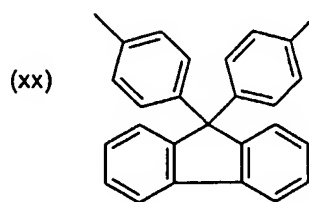
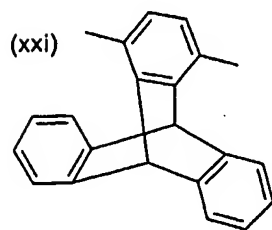
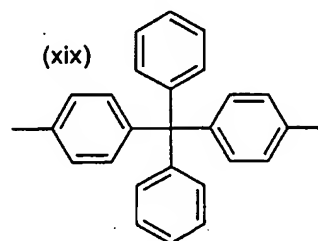
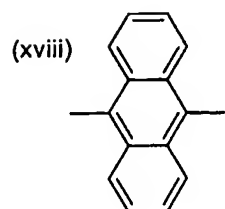
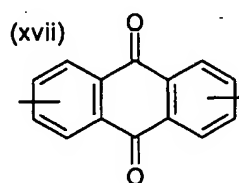
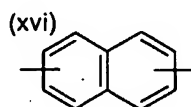
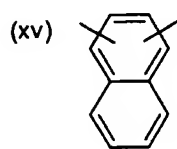
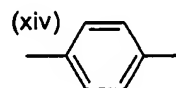
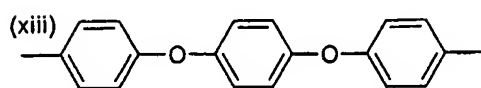
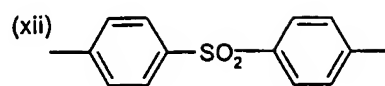
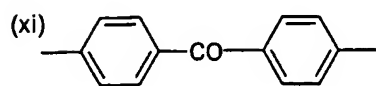
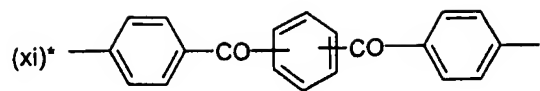
- 15 or a random or block copolymer of at least two different units of IV* and/or V*, wherein A, B, C, and D independently represent 0 or 1 and E, E', G, Ar, m, r, s, t, v, w and z are as described in any statement herein.

- 20 Preferably, m is in the range 0-3, more preferably 0-2, especially 0-1. Preferably, r is in the range 0-3, more

preferably 0-2, especially 0-1. Preferably t is in the range 0-3, more preferably 0-2, especially 0-1. Preferably, s is 0 or 1. Preferably v is 0 or 1. Preferably, w is 0 or 1. Preferably z is 0 or 1.

5

Preferably Ar is selected from the following moieties (xi)* and (xi) to (xxi):



In (xi)*, the middle phenyl may be 1,4- or 1,3-substituted.

Preferably, (xv) is selected from a 1,2-, 1,3-, or a 1,5- moiety; (xvi) is selected from a 1,6-, 2,3-, 2,6- or a 2,7- moiety; and (xvii) is selected from a 1,2-, 1,4-, 1,5-, 1,8- or a 2,6- moiety.

5

One preferred class of semi-crystalline polymers may include at least some ketone moieties in the polymeric chain. Thus, in this case, suitably, a polymer of the first aspect includes moieties of formula I and/or II.

10

One preferred class of semi-crystalline polymers does not include any moieties of formula III, but suitably only includes moieties of formulae I and/or II. Where said polymer is a homopolymer or random or block copolymer as described, said homopolymer or copolymer suitably includes a repeat unit of general formula IV. Such a polymer may, in some embodiments, not include any repeat unit of general formula V.

20 Suitable moieties Ar are moieties (i)*, (i), (ii), (iv) and (v) and, of these, moieties (i)*, (i), (ii) and (iv) are preferred. Preferred moieties Ar are moieties (xi)*, (xi), (xii), (xiv), (xv) and (xvi) and, of these, moieties (xi)*, (xi), (xii) and (xiv) are especially preferred.

25 Another preferred moiety is moiety (v), especially, moiety (xvi). In relation, in particular to the alternative semi-crystalline polymers comprising units IV* and/or V*, preferred Ar moieties are (v) and, especially, (xvi).

30 Preferred semi-crystalline polymers include an electron-rich, relatively non-deactivated, easily sulphonatable unit, for example a multi-phenylene moiety or a fused-rings aromatic moiety, such as naphthalene. Such an

easy to sulphonate unit may be sulphonated under relatively mild conditions to introduce two sulphonate groups per unit. Thus, preferred polymers may have at least 10π electrons in a delocalized aromatic moiety. The number of π electrons may be 12 or less. Preferred polymers include a biphenylene moiety. Other preferred polymers include a naphthalene moiety. Preferred polymers include said electron rich, non-deactivated, easily sulphonatable unit bonded to two oxygen atoms. Especially preferred polymers include a -O-biphenylene-O- moiety. Other especially preferred polymers include a -O-naphthalene-O- moiety.

Preferred semi-crystalline polymers include a first type of moiety which is relatively difficult to sulphonate and a second type of moiety which is relatively easy to sulphonate. For example, said second moiety may be sulphonatable using the relatively mild method described in Example 4 hereinafter, whereas the first moiety may be substantially non-sulphonatable in such a method. The use of the method of Example 4 may be advantageous over currently used methods which use oleum. A preferred second said moiety includes a moiety $-\text{Ph}_n-$ wherein n is an integer of at least 2. Said moiety is preferably bound to at least one ether oxygen. Especially preferred is the case wherein said moiety is $-\text{O-Ph}_n-\text{O-}$ where said ether groups are para to the Ph-Ph bond.

Preferred semi-crystalline polymers are copolymers comprising, preferably consisting essentially of, a first repeat unit which is selected from the following:

(a) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a

moiety of structure (iv), m and s represent zero, w represents 1 and A and B represent 1;

(b) a unit of formula IV wherein E represents an oxygen
5 atom, E' represents a direct link, Ar represents a moiety of structure (i), m represents zero, A represents 1, B represents zero;

(c) a unit of formula V wherein E and E' represent
10 oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m and v represent zero, z represents 1 and C and D represent 1;

(d) a unit of formula V wherein E represents an oxygen
15 atom, E' represents a direct link, Ar represents a moiety of structure (ii), m represents 0, C represents 1, D represents 0; or

(e) a unit of formula V wherein E and E' represents an
20 oxygen atom, Ar represents a structure (i), m represents 0, C represents 1, Z represents 1, G represents a direct link, v represents 0 and D represents 1;

(aa) a unit of formula IV wherein E represents an
25 oxygen atom E' represents a direct link, Ar represents a structure (i)*, m represents 0, A represents 1, B represents 0;

(bb) a unit of formula IV wherein E and E' represent
30 oxygen atoms, Ar represents a structure (iv), m and w represent 0, G represents a direct link, s and r represent 1, A and B represent 1;

(cc) a unit of formula IV wherein E and E' represent oxygen atoms, Ar represents a structure (i), m and w represent 0, G represents a direct link, s and r represent 1, A and B represent 1;

5

and a second repeat unit which is selected from the following:

(f) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m represents 1, w represents 1, s represents zero, A and B represent 1;

(g) a unit of formula IV wherein E represents an oxygen atom, E' is a direct link, G represents a direct link, Ar represents a moiety of structure (iv), m and s represent zero, w represent 1, A and B represent 1;

(h) a unit of formula V wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m represents 1, z represents 1, v represents 0, C and D represent 1; and

(i) a unit of formula V wherein E represents an oxygen atom, E' represents a direct link, G represents a direct link, Ar represents a moiety of structure (iv), m and v represent zero, z represents 1, C and D represent 1;

(j) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (v), m represents 0, w represents 1, s represents 0, A and B represent 1;

(k) a unit of formula V wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (v), m represents 0, z represents 1, v represents 0, C and D represent 1.

5

Preferred semi-crystalline polymers for some situations may comprise first units selected from (a), (b), (c) and (e) and second units selected from (f), (g), (h) or (i). A polymer comprising units (d) and (h) may also be preferred.

10 In some situations, first units may be selected from (aa), (bb) and (cc) and second units may be selected from (f), (g), (h) or (i).

More preferred semi-crystalline polymers are copolymers
15 having a first repeat unit selected from those described above, especially repeat units (b), (d) or (e) in combination with a second repeat unit selected from units (f) or (h). Other particularly preferred polymers are copolymers having a first repeat unit selected from (aa)
20 and (bb) in combination with a second repeat unit selected from (f) or (h).

In some situations, a difficult to sulphonate unit may include at least one relatively strongly electron-
25 withdrawing group (e.g. -CO- or -SO₂- group) bonded to a phenyl group. Such a unit will be more difficult to sulphonate compared to, for example, a unit having a phenyl group not bonded to such a strongly electron-withdrawing group. Thus, in this case, a copolymer comprising a unit
30 (a) or (c) in combination with difficult to sulphonate units as described may be prepared. Preferred copolymers of this type may comprise first (difficult to sulphonate) repeat unit(s) of formula (b) and/or (d) together with

second relatively easy to sulphonate) unit(s) of formula (a) and/or (c). Especially preferred copolymers comprise, preferably consist essentially of a first (difficult to sulphonate) repeat unit of formula (b) or (d) together with
5 a second (relatively easy to sulphonate) unit of formula (a) or (c).

Preferred first materials having repeat unit(s) of formulae IV* and V* may include: a unit of formula IV*
10 wherein Ar represents a moiety of structure (v), E represents a direct link, E' represents an oxygen atom, G represents a direct link, w, s and m represent 0, A and B represent 1; and/or a repeat unit of formula V* wherein Ar represents a moiety of structure (v), E represents a direct
15 link, E' represents an oxygen atom, G represents a direct link, z, v and m represent 0, C and D represent 1.

Said first materials having repeat units IV* and V* may include any of repeat units (a) to (i) (and/or units (aa),
20 (bb) and (cc)) described above.

In some situations, semi-crystalline polymers which include at least one repeat unit of formula IV or formula IV* may be preferred.
25

Copolymers may be prepared having one or more first repeat units and one or more of said second repeat units.

Where said first material is a copolymer as described,
30 the mole% of co-monomer units, for example said first and second repeat units described above, may be varied to vary the solubility of the polymer in solvents, for example in organic solvents which may be used in the preparation of

films and/or membranes from the polymers and/or in other solvents, especially water. Also, the mole% of co-monomer units may be varied to vary the level of crystallinity and/or crystallisability. For homopolymers, the level of
5 crystallinity and/or crystallisability may be determined by the level of functionalisation with ion-exchange sites.

Thus, if, for example, polyetherketone or polyetheretherketone are highly sulphonated (to provide
10 high conductivity or high solubility in solvents), then the polymers will have no or low crystallinity. However, if the polymers are less highly sulphonated, the polymers may be semi-crystalline and, therefore, exhibit the advantageous properties described herein when used in
15 DMFCs.

Preferred semi-crystalline polymers have a solubility of at least 10% w/v, preferably a solubility in the range 10 to 30 %w/v in a polar aprotic solvent, for example NMP,
20 DMSO or DMF. Preferred polymers are substantially insoluble in boiling water.

First units of the type described above (with the exception of units (a) and (c) in the situation described
25 above) may be relatively difficult to sulphonate, whereas second units of the type described may be easier to sulphonate.

Where a phenyl moiety is sulphonated, it may only be
30 mono-sulphonated. However, in some situations it may be possible to effect bi- or multi-sulphonation.

In general terms, where a said semi-crystalline polymer includes a -O-phenyl-O- moiety, up to 100 mole% of the phenyl moieties may be sulphonated. Where a said first material includes a -O-biphenylene-O- moiety, up to 100 mole% of the phenyl moieties may be sulphonated. It is believed to be possible to sulphonate relatively easily -O-(phenyl)_n-O- moieties wherein n is an integer, suitably 1-3, at up to 100 mole%. Moieties of formula -O-(phenyl)_n-CO- or -O-(phenyl)_n-SO₂- may also be sulphonated at up to 100 mole% but more vigorous conditions may be required. Moieties of formulae -CO-(phenyl)_n-CO- and -SO₂-(phenyl)_n-SO₂- are more difficult to sulphonate and may be sulphonated to a level less than 100 mole% or not at all under some sulphonation conditions.

15

The glass transition temperature (T_g) of said first material may be at least 144°C, suitably at least 150°C, preferably at least 154°C, more preferably at least 160°C, especially at least 164°C. In some cases, the T_g may be at least 170°C, or at least 190°C or greater than 250°C or even 300°C.

Said semi-crystalline polymer may have an inherent viscosity (IV) of at least 0.1, suitably at least 0.3, preferably at least 0.4, more preferably at least 0.6, especially at least 0.7 (which corresponds to a reduced viscosity (RV) of least 0.8) wherein RV is measured at 25°C on a solution of the polymer in concentrated sulphuric acid of density 1.84gcm⁻³, said solution containing 1g of polymer per 100cm⁻³ of solution. IV is measured at 25°C on a solution of polymer in concentrated sulphuric acid of density 1.84gcm⁻³, said solution containing 0.1g of polymer per 100cm³ of solution.

The measurements of both RV and IV both suitably employ a viscometer having a solvent flow time of approximately 2 minutes.

5

The main peak of the melting endotherm (T_m) for said semi-crystalline polymer may be at least 300°C.

Said semi-crystalline polymer may include a
10 crystalline or crystallisable unit which is of general formula IV or IV* as described above, provided said unit is crystallisable. Suitably, to be crystallisable, said unit does not include any Ar group of formula (ii), (viii), (ix) or (x). More preferably, it may also not
15 include an Ar group of formula (v), (vi) or (vii). Preferred Ar groups consist of one or more phenyl groups in combination with one or more carbonyl and/or ether groups.

20 In one embodiment, said semi-crystalline polymer is a copolymer comprising:

a first unit which includes an ion-exchange site;
a second crystalline unit; and, optionally,
25 a third unit which is amorphous.

Suitably, "A*" represents the mole% of said first unit in said copolymer; "B*" represents the mole % of said second unit; and "C*" represents the mole % of said third
30 unit.

Suitably, A* is at least 15, preferably at least 20, more preferably at least 25, especially at least 30. It

may be less than 70, preferably less than 60, more preferably less than 50. Preferably, A* is in the range 25-60.

5 B* may be at least 5. Suitably, B* is at least 15, preferably at least 25, more preferably at least 30, especially at least 35. It may be less than 70, preferably less than 60, more preferably less than 55. Preferably, B* is in the range 5-70.

10

Suitably, C* is at least 5, preferably at least 7.5, preferably at least 10, especially at least 12.5. In some cases C* may be at least 25. C* may be less than 70, preferably less than 60, more preferably less than 55. In
15 other cases, C* may be less than 30, preferably less than 25, more preferably less than 20, especially 15 or less. Preferably, C* is in the range 5 to 70.

Said first unit is preferably a repeat unit which
20 suitably includes aromatic group containing moieties linked by -SO₂- and/or -CO- and/or -Q- groups, where Q is O or S. Because said first unit includes ion-exchange sites, for example, sulphonate groups, it may not be crystalline, but will be amorphous.

25

Said second unit is preferably a repeat unit which suitably includes aromatic group containing moieties linked by -CO- and/or -Q- groups, where Q is as described above. The second unit preferably does not include -SO₂-
30 groups since such would tend to render the unit amorphous.

Said third unit is preferably a repeat unit which suitably includes aromatic group containing moieties

linked by -SO₂- and/or -CO- and/or -Q- groups, where Q is as described above provided, however, that said third unit suitably includes a means to render it amorphous (hereinafter said "amorphous means") and/or not
5 crystallisable with polyarylether ketones or polyarylthioether ketones and/or not crystallisable with the second unit described above.

Said third unit may comprise a fourth unit which is of
10 formula -Q-Z-Q- wherein Z represents said aromatic group containing moiety, wherein said fourth unit is not symmetrical about an imaginary line which passes through the two -Q- moieties provided, however, that said fourth unit is not dihydroxybenzophenone substituted by groups Q
15 at the 4- and 4'- positions (since such a benzophenone acts in the manner of a symmetrical moiety by virtue of the carbonyl group being substantially similar to an ether group thereby allowing the carbonyl group to be interchanged with an ether group in a polyaryletherketone
20 crystal lattice). Said third unit, for example moiety Z, may include a bulky group.

Said semi-crystalline copolymer may include:

25 - a first unit of formula IV, V, IV* or V* as described above, wherein said first unit is functionalised to provide ion-exchange sites.

- a second crystalline unit which is of general formula
30 IV or IV* as described above, provided said unit is crystallisable. Suitably, to be crystallisable, said second unit does not include any Ar group of formula (ii), (viii), (ix) or (x). More preferably, it may also not

include an Ar group of formula (v), (vi) or (vii). Preferred Ar groups consist of one or more phenyl groups in combination with one or more carbonyl and/or ether groups.

- 5 - a third unit which is of general formula IV, IV*, V or V*, provided, however, that said unit includes at least some moieties whose shape and/or conformation is/are incompatible with the crystalline conformation of said second crystalline unit so that said third unit is
10 amorphous. Preferably, said third unit includes an -SO₂- moiety; a bulky group or a moiety which is not symmetrical as described above.

Preferred first units may be -ether-phenyl-ketone-phenyl, -ether-phenyl-ketone-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl, -ether-biphenyl-ether-phenyl-ketone-phenyl, ether-phenyl-ether-phenyl-ketone-phenyl,, ether-naphthalene-ether-phenyl-ketone-phenyl), ether-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl, -ether-diphenyl-ether-phenyl-sulphone-phenyl- and -ether-phenyl-ether-phenyl-sulphone-phenyl, suitably functionalised with ion-exchange sites. Another preferred first unit is ether-phenyl-sulphone-phenyl. Preferred second units may be ether-phenyl-ketone-phenyl-ketone-phenyl-, ether-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl-, ether-phenyl-ether-phenyl-ketone-phenyl-, ether-phenyl-ketone-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl and ether-biphenyl-ether-phenyl-ketone-phenyl-. Preferred third units may be ether-phenyl-sulphone-phenyl and ether-phenyl-ether-phenyl-sulphone-phenyl. Another preferred third unit may be a - 1,3-dioxy-4-(phenylcarbonyl) phenyl moiety derived from 2,4-DHB as herein defined.

15
20
25
30

In said copolymer, the mole% of co-monomer units, for example said first, second and third repeat units described above, may be varied to vary the solubility of the polymer
5 in solvents, for example in solvents which may be used in the preparation of films and/or membranes from the polymers and/or in other solvents, especially water.

Preferred polymers suitably have a solubility of at
10 least 4% w/w in a polar aprotic solvent, for example NMP, DMSO or DMF. Preferred polymers are substantially insoluble in boiling water.

The semi-crystalline polymer preferably includes less
15 than 10%, more preferably less than 5%, especially less than 1% of cross-linked $-SO_3H$ groups i.e. $-SO_2-$ groups formed by cross-linking two $-SO_3H$ groups. Preferably, the semi-crystalline polymer includes no cross-linked $-SO_3H$ groups and no cross-linking in the polymer at all.

20

Said semi-crystalline polymer could be a component of a blend of polymers - that is, the PEM may comprise a blend of polymers. In such a blend, said semi-crystalline polymer suitably comprises at least 50wt%, preferably at
25 least 80wt%, more preferably at least 95wt%. Preferably, said semi-crystalline polymer is not a component of an acid-base blend. Preferably, said semi-crystalline polymer is not blended with a nitrogen-containing basic polymer. Preferably, said semi-crystalline copolymer is
30 not a component of a blend.

Said semi-crystalline polymer could be a component of a composite membrane in conjunction with a support

material which may be non-conducting as described in PCT/GB00/03449 or conducting as described in GB0031209.0.

Said PEM may have a dimension in a first direction of
5 at least 1cm. The dimension of the PEM in a second direction, perpendicular to the first direction, may also be at least 1cm. Where the PEM is circular, the diameter may be at least 1cm. In some cases, for example for vehicle applications, the dimension(s) in the first and/or
10 second direction(s) may be at least 10cm or at least 20cm. The dimension(s) in the first and second direction(s) is/are suitably less than 100cm, preferably less than 50cm, more preferably less than 35cm.

15 Said semi-crystalline polymer may comprise a film, suitably having a thickness of less than 1mm, preferably less than 0.5mm, more preferably less than 0.1mm, especially less than 0.05mm. The film may have a thickness of at least 5 μ m.

20

Said polymer electrolyte membrane may comprise one or more layers wherein, suitably, at least one layer comprises a film of said semi-crystalline polymer. Said membrane may have a thickness of at least 5 μ m and,
25 suitably, less than 1mm, preferably less than 0.5mm, more preferably less than 0.1mm, especially less than 0.05mm.

Preferably, the identity of the semi-crystalline polymer at a surface or interface thereof is the same as
30 the bulk of the polymer -that is, said semi-crystalline polymer is preferably not surface modified.

Said FCDF preferably includes an anode and a cathode suitably operatively connected to said semi-crystalline polymer of the PEM. Said anode and/or said cathode suitably include an ion-conducting polymer which may be as
5 described in any statement herein. Preferably said semi-crystalline polymer of the PEM and that of the anode and/or cathode are made of the same polymeric material.

Said FCDF preferably includes a fuel supply
10 receptacle. A pump is preferably provided for pumping the fuel to an anode chamber on one side of the PEM. Means is preferably provided for supplying an oxidising gas (e.g. air or oxygen) to the opposite side of the PEM. Said FCDF (e.g. DMFC) may be as described in the introduction of the
15 present specification.

According to a second aspect of the invention, there is provided a plurality of fuel cells as described according to the first aspect. The fuel cells are
20 preferably substantially identical to one another and are preferably provided in a stack in series. Greater than 50 or even greater than 100 of said fuel cells may be provided in a said stack.

25 Said plurality of fuel cells may together include more than 0.1m^2 , suitably more than 0.5m^2 , preferably more than 1m^2 , more preferably more than 5m^2 of said semi-crystalline polymer. The amount of said polymer may be less than 100m^2 .

30

According to a third aspect of the present invention, there is provided a method of operating a FCDF, for example a DMFC, wherein the FCDF is as described according

to the first aspect, the method comprising supplying a fuel to one side of the polymer electrolyte membrane (PEM). Suitably, an oxidizing gas, for example oxygen or air, is supplied to the other side of the PEM. The pressure of the gas may be at least 5 psig, preferably at least 10 psig. The pressure may be less than 50 psig, preferably less than 40 psig, more preferably less than 30 psig.

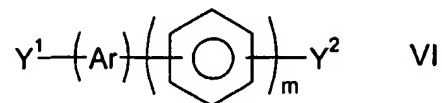
Said fuel is preferably a carbon-containing fuel. Said fuel preferably includes an alkyl moiety, for example a C_aH_{2a+1} - moiety where a is an integer suitably in the range 1-20 preferably 1-10, especially 1-5. Said fuel is preferably an alcohol. An especially preferred fuel is methanol.

According to a fourth aspect of the present invention, there is provided a method of manufacturing a FCDF, for example a DMFC having low fuel (e.g. methanol) cross-over, the method comprising incorporating a semi-crystalline polymer into a polymer electrolyte membrane of the fuel cell.

According to a fifth aspect of the present invention, there is provided the use of a semi-crystalline polymer for the manufacture of a polymer electrolyte membrane of a fuel cell, for reducing the level of fuel (e.g. methanol) cross-over in the cell, suitably compared to level of cross-over resulting from using similar amorphous polymers of the same conductivity.

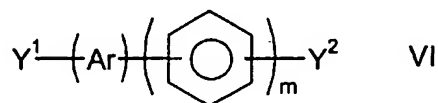
Polymers having units I, II, III, IV, IV*, V and/or V* described above may be prepared by:

(a) polycondensing a compound of general formula



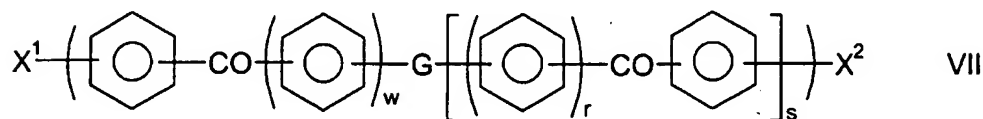
with itself wherein Y^1 represents a halogen atom or a
 5 group $-EH$ and Y^2 represents a halogen atom or, if Y^1
 represents a halogen atom, Y^2 represents a group $E'H$; or

(b) polycondensing a compound of general formula

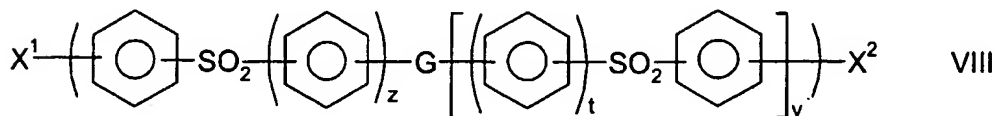


with a compound of formula

10



and/or with a compound of formula



15 wherein Y^1 represents a halogen atom or a group $-EH$ (or
 $-E'H$ if appropriate) and X^1 represents the other one of a
 halogen atom or group $-EH$ (or $-E'H$ if appropriate) and Y^2
 represents a halogen atom or a group $-E'H$ and X^2 represents

the other one of a halogen atom or a group -E'H (or -EH if appropriate).

(c) optionally copolymerizing a product of a process as
5 described in paragraph (a) with a product of a process as
described in paragraph (b);

wherein the phenyl moieties of units VI, VII and/or
VIII are optionally substituted; the compounds VI, VII
10 and/or VIII are optionally sulphonated; and Ar, m, w, r, s,
z, t, v, G, E and E' are as described above except that E
and E' do not represent a direct link;

the process also optionally comprising sulphonating
15 and/or cross-linking a product of the reaction described in
paragraphs (a), (b) and/or (c) to prepare said polymer.

In some situations, the polymer prepared, more
particularly phenyl groups thereof, may be optionally
20 substituted with the groups hereinabove described after
polymer formation.

Preferably, where Y^1 , Y^2 , X^1 and/or X^2 represent a
halogen, especially a fluorine, atom, an activating group,
25 especially a carbonyl or sulphone group, is arranged ortho-
or para- to the halogen atom.

Preferred halogen atoms are fluorine and chlorine
atoms, with fluorine atoms being especially preferred.
30 Preferably, halogen atoms are arranged meta- or para- to
activating groups, especially carbonyl groups.

Where the process described in paragraph (a) is carried out, preferably one of Y^1 and Y^2 represents a fluorine atom and the other represents an hydroxy group. More preferably in this case, Y^1 represents a fluorine atom and Y^2 represents an hydroxy group. Advantageously, the process described in paragraph (a) may be used when Ar represents a moiety of structure (i) and m represents 1.

When a process described in paragraph (b) is carried out, preferably, Y^1 and Y^2 each represent an hydroxy group. Preferably, X^1 and X^2 each represent a halogen atom, suitably the same halogen atom.

The polycondensation reaction described is suitably carried out in the presence of a base, especially an alkali metal carbonate or bicarbonate or a mixture of such bases. Preferred bases for use in the reaction include sodium carbonate and potassium carbonate and mixtures of these.

20

The identity and/or properties of the polymers prepared in a polycondensation reaction described may be varied according to the reaction profile, the identity of the base used, the temperature of the polymerisation, the solvent(s) used and the time of the polymerisation. Also, the molecular weight of a polymer prepared controlled by using an excess of halogen or hydroxy reactants, the excess being, for example, in the range 0.1 to 5.0 mole%.

In a polymer prepared in a said polycondensation reaction involving compounds of general formula VI, VII, and VIII, moieties of general formula VI, VII, and VIII (excluding end groups Y^1 , Y^2 , X^1 and X^2) may be present in

regular succession (that is, with single units of one said moiety, separated by single units of another said moiety or moieties), or semi-regular succession (that is, with single units of one said moiety separated by strings of another moiety or moieties which are not all of the same length) or in irregular succession (that is, with at least some multiple units of one moiety separated by strings of other moieties that may or may not be of equal lengths). The moieties described are suitably linked through ether or thioether groups.

Also, moieties in compounds VI, VII and VIII arranged between a pair of spaced apart -O- atoms and which include a -phenyl-SO₂ or -phenyl-CO- bonded to one of the -O- atoms may, in the polymer formed in the polycondensation reaction, be present in regular succession, semi-regular succession or in irregular succession, as described previously.

In any sampled polymer, the chains that make up the polymer may be equal or may differ in regularity from one another, either as a result of synthesis conditions or of deliberate blending of separately made batches of polymer.

Compounds of general formula VI, VII and VIII are commercially available (eg from Aldrich U.K.) and/or may be prepared by standard techniques, generally involving Friedel-Crafts reactions, followed by appropriate derivatisation of functional groups. The preparations of some of the monomers described herein are described in P M Hergenrother, B J Jensen and S J Havens, Polymer 29, 358 (1988), H R Kricheldorf and U Delius, Macromolecules 22,

517 (1989) and P A Staniland, Bull. Soc. Chem. Belg., 98
(9-10), 667 (1989).

Where compounds VI, VII and/or VIII are sulphonated,
5 compounds of formulas VI, VII and/or VIII which are not
sulphonated may be prepared and such compounds may be
sulphonated prior to said polycondensation reaction.

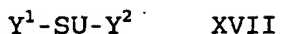
Sulphonation as described herein may be carried out in
10 concentrated sulphuric acid (suitably at least 96% w/w,
preferably at least 97%w/w, more preferably at least
98%w/w; and preferably less than 98.5%w/w) at an elevated
temperature. For example, dried polymer may be contacted
with sulphuric acid and heated with stirring at a
15 temperature of greater than 40°C, preferably greater than
55°C, for at least one hour, preferably at least two hours,
more preferably about three hours. The desired product may
be caused to precipitate, suitably by contact with cooled
water, and isolated by standard techniques. Sulphonation
20 may also be effected as described in US5362836 and/or
EP0041780.

Where a semi-crystalline copolymer is prepared
comprising a first unit, a second crystalline unit and a
25 third amorphous unit as described above, the process may
comprise:

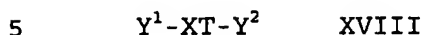
polycondensing a compound of formula

30 X^1-BM-X^2 XVI

with a compound of formula



and with a compound of formula



and with a compound of formula



10

thereby to prepare a copolymer, wherein Y^1 represents a halogen atom or a group -EH (or -E'H if appropriate) and X^1 represents the other one of a halogen atom or group -EH (or -E'H if appropriate), Y^2 represents a halogen atom or a group -E'H and X^2 represents the other one of a halogen atom or a group -E'H (or -EH if appropriate) and Z^1 and Z^2 represent a halogen atom or a group -EH (or E'H if appropriate);

20 and wherein BM represents part of a base monomer, SU represents part of a moiety which is functionalised or can be functionalised (suitably independently of other moieties in the copolymer) to provide ion-exchange sites, XT represents a part of a crystalline or crystallisable moiety
25 and AM represents part of an amorphous moiety.

The polycondensation reaction may be carried out as described above.

30 Advantageously, a crystallisable or semi-crystalline polymer for use in FCDFs as described herein may be treated as described in GB0031207.4 with a means to increase its crystallinity (hereinafter "crystallinity

increasing means"). The level of crystallinity may be at least 1%, suitably at least 5%, preferably at least 10%, more preferably at least 15%, especially at least 20% greater after treatment than the level before treatment.

5

Said crystallinity increasing means may comprise heating said crystallisable or semi-crystalline polymer, suitably when in a substantially dry state. Preferably, said polymer is heated at a temperature greater than its
10 Tg, suitably under an inert, e.g. a nitrogen, atmosphere, for at least 0.5 minutes and less than 30 minutes. Alternatively and/or additionally, said crystallinity increasing means may comprise treatment of said first ion-conducting polymeric material with a crystallinity
15 increasing solvent. Preferred such solvents are polar aprotic solvents and may include acetone, dimethylacetamide (DMA), dimethylformamide (DMF), tetrahydrofuran (THF) and dichloromethane. After use of a crystallinity increasing solvent, especially acetone or dichloromethane, there is
20 preferably a further step which involves removal of the solvent, for example by washing with acetone.

Details on the preparation of polymers and processes for the preparation of membranes therefrom are provided in
25 WO00/15691, PCT/GB00/03449, GB0031209.0, GB0031208.2 and GB0031207.4 and the contents of the aforesaid are incorporated herein by reference.

Any feature of any aspect of any invention or
30 embodiment described herein may be combined with any feature of any aspect of any other invention or embodiment described herein.

Specific embodiments of the invention will now be described, by way of example, with reference to Figure 1 which is a schematic representation of a Direct Methanol Fuel Cell (DMFC).

5

The following abbreviations are used hereinafter

BP - 4,4'-dihydroxybiphenyl
DHB - 4,4'-dihydroxybenzophenone
10 Bis-S - 4,4'-dihydroxydiphenylsulphone
2,4-DHB is 2,4-dihydroxybenzophenone
BDF - 4,4'-difluorobenzophenone

Unless otherwise stated, all chemicals referred to
15 hereinafter were used as received from Sigma-Aldrich Chemical Company, Dorset, U.K.

The examples describe the preparation of polymers which are crystallisable and/or semi-crystalline and can,
20 therefore, be used as polymer electrolyte membranes to reduce methanol cross over in DMFCs. Various examples of such polymers are included to illustrate how the level of crystallinity may be adjusted. For example, crystallinity can be reduced by incorporation of moieties derived from
25 2,4-DHB and/or Bis-S in the polymer chain (see Examples 1b-d, 2b-2g, 3a-3c and 7a to 7f). Additionally, treatments are described (see Examples 6, 9, 14, 16 and 18) whereby polymers can be treated to increase their crystallinity and/or boiling water resistance. Examples of
30 amorphous polymers are included for comparison with the crystallisable/crystalline polymers.

A comparison of methanol cross-over for polymer membranes made out of the known material Nafion, an amorphous polyaryletherketone sulphone and a semi-crystalline polymer are also included to illustrate the advantages of using semi-crystalline polymers described.

Example 1a

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (37.24g, 0.20 mole) 4,4'-dihydroxybenzophenone (42.84g, 0.20 mole), and diphenysulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was added. The temperature was raised gradually to 330°C over 3 hours then maintained for 1 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.48 kNsm⁻².

Example 1b

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (37.24g, 0.20 mole), 4,4'-dihydroxydiphenylsulphone (10.01g, 0.04 mole), 4,4'-

dihydroxybenzophenone (34.28, 0.16 mole) and diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost
5 colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was added. The temperature was raised gradually to 320°C over 3 hours then maintained for 1.5 hours.

10 The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.34 kNsm⁻².

15 Example 1c

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g,
20 0.408 mole), 4,4'-dihydroxybiphenyl (37.24g, 0.20 mole), 4,4'-dihydroxydiphenylsulphone (15.02g, 0.06 mole), 4,4'-dihydroxybenzophenone (29.99g, 0.14 mole) and diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen
25 blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was added. The temperature was raised gradually to 320°C over 3 hours then maintained for 1 hours.

30

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was

dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.42 kNsm⁻².

Example 1d (comparative)

5

A 700 flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (37.24g, 0.20 mole)
10 4,4'-dihydroxydiphenylsulphone (50.05g, 0.20 mole), and diphenysulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen
15 blanket, dried sodium carbonate (42.44g, 0.4 mole) and potassium carbonate (1.11g, 0.008 mole) was added. The temperature was raised gradually to 315°C over 3 hours then maintained for 30 minutes.

20 The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.43 kNsm⁻².

25 Example 2a

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (24.83g, 0.133 mole)
30 4,4'-dihydroxybenzophenone (57.41g, 0.268 mole), and diphenysulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen

blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was added. The temperature was raised gradually to 330°C over
5 3 hours then maintained for 1 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt
10 viscosity at 400°C, 1000sec⁻¹ of 0.54 kNsm⁻².

Example 2b

A 700ml flanged flask fitted with a ground glass
15 Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (24.83g, 0.133 mole), 4,4'-dihydroxydiphenylsulphone (13.35g, 0.053 mole), 4,4'-dihydroxybenzophenone (45.7g, 0.213 mole) and
20 diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was
25 added. The temperature was raised gradually to 320°C over 3 hours then maintained for 1.5 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was
30 dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.37 kNsm⁻².

Examples 2c-2f and 2g(comparative)

The polymerisation procedure of Example 2b was followed, for 2c-2f except that copolymers were prepared by varying the mole ratios of the hydroxy-containing reactants. The polymerisation procedure for 2g is described below.

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (24.83g, 0.133 mole) 4,4'-dihydroxydiphenylsulphone (66.73g, 0.267 mole), and diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (42.44g, 0.4 mole) and potassium carbonate (1.11g, 0.008 mole) were added. The temperature was raised gradually to 315°C over 3 hours then maintained for 0.5 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.62 kNsm⁻².

A summary of the mole ratios and MVs are detailed in the Table below. Example 2g is an amorphous equivalent of the other polymers.

Example	Polymer composition (mole ratio)				MV (kNsm ⁻²)
	BDF	BP	DHB	Bis-S	
2a	1.02	0.33	0.67	-	0.54
2b	1.02	0.33	0.536	0.133	0.37
2c	1.02	0.33	0.402	0.268	0.47
2d	1.02	0.33	0.335	0.335	0.48
2e	1.02	0.33	0.268	0.402	0.48
2f	1.02	0.33	0.133	0.536	0.53
2g	1.02	0.33	-	0.67	0.62

Example 3a

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (24.83g, 0.133 mole), 2,4-dihydroxybenzophenone (11.42g, 0.053 mole), 4,4'-dihydroxybenzophenone (45.7g, 0.213 mole) and diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was added. The temperature was raised gradually to 320°C over 3 hours then maintained for 1.5 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.80 kNsm⁻².

Example 3b and 3c

The polymerisation procedure of Example 3a was followed except that copolymers were prepared with different mole ratios of the hydroxy-containing reactants. A summary of the mole ratios and MVs for Examples 3a-3c are detailed in the Table below.

Example	Polymer Composition (mole ratio)				MV (kNsm ⁻²)
	BDF	BP	4,4'-DHB	2,4-DHB	
3a	1.02	0.33	0.533	0.133	0.70
3b	1.02	0.33	0.402	0.268	0.38
3c	1.02	0.33	0.133	0.533	0.47

Example 4 (General Sulphonation Procedure)

10

The polymers of Examples 1-3 were sulphonated by stirring each polymer in 98% sulphuric acid (3.84g polymer/100g sulphuric acid) for 21 hours at 50°C. Thereafter, the reaction solution was allowed to drip into stirred deionised water. Sulphonated polymer precipitated as free-flowing beads. Recovery was by filtration, followed by washing with deionised water until the pH was neutral and subsequent drying. In general, titration confirmed that 100 mole% of the biphenyl units had sulphonated, giving one sulphonic acid group, ortho to the ether linkage, on each of the two aromatic rings comprising the biphenyl unit. Also, it was confirmed that the naphthalene unit was sulphonated on both aromatic rings in the polymers of Example 3a-3c.

25

Example 5a (Membrane Fabrication)

Membranes were produced from the polymers from Examples 1 to 3 after sulphonation as described in Example 4 by dissolving respective polymers in N-methylpyrrolidone (NMP). The polymers were dissolved at 80°C at their maximum concentration as shown in the Table below. In one example, a 50:50 w/w blend of the polymers described in Examples 2e and 2f, sulphonated as described in Example 4, was used to prepare a membrane.

The homogeneous solutions were cast onto clean glass plates and then drawn down to give 400 micron films, using a Gardner Knife. The solvent was then evaporated at 100°C under vacuum for 24 hours.

Example 5b (Boiling water uptake)

The following general procedure was followed to determine the Boiling Water Uptake.

5cm x 5cm x 50 microns (unless otherwise stated) samples of membranes were separately immersed in boiling deionised water (500ml) for 60 mins, removed and dried quickly with lint-free paper to remove surface water, weighed, dried in an oven at 50°C for 1 day, allowed to cool to ambient temperature in a desiccator then weighed quickly. The % water-uptake was calculated as described below:

$$\% \text{ Water-uptake} = \frac{\text{Wet Weight} - \text{Dry Weight}}{\text{Dry Weight}} \times 100$$

Results for membranes assessed are provided in the Table presented in Example 6.

Example 6 - Post treatment of the membranes prepared from sulphonated polymers 1-3 with acetone

5cm x 5cm x 40 microns samples of membranes prepared from polymers of Examples 1-3 after sulphonation and fabrication as described in Examples 4 and 5a were separately immersed in refluxing acetone (100ml) for 60 mins, removed and dried in an oven at 50°C for 1 day, immersed in boiling deionised water (500ml) for 60 mins, removed and dried quickly with lint-free paper to remove surface water, weighed, dried in an oven at 50°C for 1 day, allowed to cool to ambient temperature in a desiccator then weighed quickly. The % water uptake of each membrane was as calculated previously and summarised in the Table below. Whilst the sulphonated polymer from Example 2a was insoluble in NMP at 80°C (the temperature at which membranes were fabricated - see Example 5a), the polymer was found to be soluble in NMP at higher temperatures.

20

Sulphonated polymer from Example*	Theoretical EW	Measured EW (by titration)	Concentration in NMP (%w/w)	Boiling Water Uptake before acetone treatment (%)	Boiling Water Uptake after acetone treatment (%)
1a	458	472	4	237	116
1b	462	483	7.5	300	125
1c	464		10	320	152
1d	476	492	15	Soluble	Soluble
2a	654	674	Insoluble	-	
2b	657	670	5	69	61
2c	663	667	7.5	77	48
2d	670	671	7.5	81	57
2e	676	685	10	90	63
2f	683	660	15	172	100

Sulphonated polymer from Example*	Theoretical EW	Measured EW (by titration)	Concentration in NMP (%w/w)	Boiling Water Uptake before acetone treatment (%)	Boiling Water Uptake after acetone treatment (%)
2g	690	663	15	165	170
3a	647	666	5	73	58
3b	655	671	10	100	69
3c	670	681	15	518	208
50:50 w/w blend of polymers from Examples 2e and 2f	680		15	128	73

* It should be appreciated that the polymers of the referenced Examples are sulphonated as described in Example 4.

Example 7a

A 700ml flanged flask fitted with a ground glass Quickfit lid, stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (89.03g, 0.408 mole), 4,4'-dihydroxybiphenyl (29.79g, 0.16 mole), 4,4'-dihydroxydiphenylsulphone (36.04g, 0.144 mole), 4,4'-dihydroxybenzophenone (20.57g, 0.096 mole) and diphenylsulphone (332g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (43.24g, 0.408 mole) was

added. The temperature was raised gradually to 320°C over 3 hours then maintained for 1.5 hours.

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.6 kNsm⁻².

Example 7b - 7e and 7f (comparative)

10

The polymerisation procedure of Example 7a was followed, except that copolymers were prepared by varying the mole ratios of the hydroxy-containing reactants. A summary of the mole ratios and the MVs are detailed in the Table below.

15

Example	Polymer composition (mole ratio)				MV (kNsm ⁻²)
	BDF	BP	DHB	Bis-S	
7a	1.02	0.4	0.24	0.36	0.6
7b	1.02	0.4	0.36	0.24	0.21
7c	1.02	0.4	0.39	0.21	0.32
7d	1.02	0.4	0.42	0.18	0.44
7e	1.02	0.4	0.6	-	0.45
7f	1.02	0.4	-	0.6	0.26

Examples 8a-f Sulphonation and subsequent dissolution of Polymers from Examples 7a-f

20

The polymers from Examples 7a-f respectively were sulphonated as described in Example 4, dissolved in NMP at 150°C, filtered through a 10 micron filter, cast on to clean glass plates and drawn down, using a Gardner Knife. The solvent was evaporated at 100°C under vacuum for 24

25

hours producing membranes of mean thickness of 40 microns. The boiling water uptake was determined as described in Example 5b. The results are detailed in the Table in Example 9.

5

Example 9 Post treatment of the membranes prepared from sulphonated polymers 8a -8f with acetone

The procedure of Example 6 was followed, except that the membranes from 8a - 8f were immersed in refluxing acetone. The % water-uptake of each membrane was calculated as described above. The results are summarised in the Table below.

Sulphonated polymer from Example	Concentration in NMP (%w/w)	Boiling Water Uptake before acetone treatment (%)	Boiling Water Uptake After acetone treatment (%)	Theoretical EW	Measured EW (by titration)
7a	15	550	130	564	564
7b	10	190	90	559	564
7c	10	135	81	558	571
7d	10	109	58	557	591
7e	8	82	69	550	572
7f	15	520	520	583	602

15

Example 10 Blends with polyethersulphone

Sulphonated polymer from Example 2e and polyethersulphone were dissolved in N-methylpyrrolidone (NMP) at concentrations shown in the Table below. The homogeneous solutions were cast onto clean glass plates and then drawn down to give 400 micron films, using a stainless steel Gardner Knife. Evaporation at 100°C under

20

vacuum for 24 hours produced membranes of mean thickness 40 microns.

The procedure of Example 6 was followed, except that
5 the membranes from above were immersed in refluxing acetone. The % water-uptake of each membrane before and after acetone treatment was calculated as described above. The results are summarised in the Table below.

Sulphonated Polymer from Example 2e %w/w	Polyether sulphone %w/w	Boiling Water Uptake before acetone treatment (%)	Boiling Water Uptake after acetone treatment (%)
15	0	102	61
14.25	0.75	125	74
13.5	1.5	105	67

10

Example 11 - Blend with polyethersulphone

The procedure of Example 10 was followed except that
sulphonated polymer from Example 7d was used instead of
15 that from Example 2e. Results for the boiling water uptake, before and after acetone treatment, are detailed in the Table below.

Sulphonated Polymer from Example 8d %w/w	Polyether Sulphone %w/w	Boiling Water Uptake before acetone treatment (%)	Boiling Water Uptake after acetone treatment (%)
15	0	109	58
14.25	0.75	84	59
13.5	1.5	74	54
12.75	2.25	69	48
12.0	3.0	49	39

Example 12

5 A 250ml 3-necked, round-bottomed fitted with a stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (11.36g, 0.052mole), 4,4'-bis(4-chlorophenylsulphonyl)biphenyl (LCDC) (25.17g, 0.05mole), 4,4'-dihydroxybiphenyl (6.21g, 10 0.0333mole), 4,4'-dihydroxybenzophenone (14.28g, 0.0667mole), and diphenylsulphone (90g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a 15 nitrogen blanket, dried sodium carbonate (10.6g, 0.1 mole) and potassium carbonate (0.28g, 0.002 mole) were added. The temperature was raised gradually to 315°C over 3 hours then maintained for 1 hour.

20 The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was

dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.18kNsm⁻².

Example 13

5

A 250ml 3-necked, round-bottomed fitted with a stirrer/stirrer guide, nitrogen inlet and outlet was charged with 4,4'-difluorobenzophenone (11.02g, 0.0505mole), 4,4'-dichlorodiphenylsulphone (14.36g, 0.05mole), 4,4'-dihydroxybiphenyl (6.21g, 0.0333mole), 4,4'-dihydroxybenzophenone (14.28g, 0.0667mole), and diphenylsulphone (83g) and purged with nitrogen for over 1 hour. The contents were then heated under a nitrogen blanket to between 140 and 150°C to form an almost colourless solution. While maintaining a nitrogen blanket, dried sodium carbonate (10.60g, 0.1 mole) and potassium carbonate (0.28g, 0.002 mole) were added. The temperature was raised gradually to 315°C over 3 hours then maintained for 140 minutes.

20

The reaction mixture was allowed to cool, milled and washed with acetone and water. The resulting polymer was dried in an air oven at 120°C. The polymer had a melt viscosity at 400°C, 1000sec⁻¹ of 0.39 kNsm⁻².

25

Example 14 Sulphonation of and subsequent dissolution, membrane fabrication and post treatment with acetone of Polymers from Examples 12 and 13.

30

The polymers from Examples 12 and 13 were sulphonated as described in Example 4 and dissolved in NMP at 15 %w/w at 80°C and room temperature respectively. The homogeneous solutions were filtered through a 10 micron filter, cast

on to clean glass plates and drawn down to give 400 micron films, using a Gardner Knife. The solvent was evaporated at 100°C under vacuum for 24 hours. The boiling water uptake was 39 and 108% for the sulphonated polymer from Example 12 and 13 respectively, determined as described in Example 5b.

The procedure of Example 6 was followed, except that the membranes from above were immersed in refluxing acetone. The boiling water-uptake was 31 and 46% for the sulphonated polymer from Example 12 and 13 respectively after post treatment.

Example 15a-15d Sulphonation of polyetherketone and subsequent dissolution

A 500ml, 3-necked, round-bottomed flask fitted with a stirrer/stirrer guide, nitrogen inlet and outlet and a thermometer was charged with 98% sulphuric acid (180g) and, while stirring, polyetherketone (PEK™ P22, Victrex plc) (20g) was added. The temperature was increased to 55°C and oleum (20% free SO₃) (120g) was added. The solution was stirred for 60 minutes at 55°C. The solution was quickly cooled to 20°C, thereafter, allowed to drip into stirred deionised water. Sulphonated polymer precipitated as free-flowing beads. Recovery was by filtration, followed by washing with deionised water until the pH was neutral and subsequent drying. By titration the Equivalent Weight was 476.

The above procedure was repeated three times except that the sulphonating solution was heated to 35°C before oleum was added and maintained at that temperature for 60

mins before being rapidly cooled to 20°C. The Equivalent Weights of the sulphonated polymers was 568, 667 and 758.

The sulphonated polyetherketone polymers described above were separately dissolved in NMP, filtered through a 10 micron filter, cast on to clean glass plates and drawn down, using a Gardner Knife. The solvent was evaporated at 100°C under vacuum for 24 hours producing membranes of mean thickness of 40 microns. The boiling water uptake was determined as described above and the results are detailed in the Table presented in Example 16.

Example 16 - Post treatment of the Membranes from 15a - 15d with acetone.

15

The procedure of Example 6 was followed, except that the membranes from Examples 15a - 15d were immersed in refluxing acetone. The % water-uptake of each membrane was calculated as described above and summarised in the Table below.

20

Sulphonated polymer from Example	EW	Boiling Water uptake before acetone treatment (%)	Boiling Water uptake after acetone Treatment (%)
15a	476	Sample broke up	124
15b	568	950	72
15c	670	370	56
15d	758	80	51

Example 17 - Sulphonation of polyetheretherketone and subsequent dissolution

A 500ml, 3-necked, round-bottomed flask fitted with a stirrer/stirrer guide, nitrogen inlet and outlet and a thermometer was charged with 98% sulphuric acid (180g). The sulphuric acid was heated under a blanket of nitrogen to 50°C. While maintaining a nitrogen blanket and stirring polyetheretherketone (PEEK™ 450P, Victrex plc) was added. The polymer dissolved and was stirred at 50°C for 90 minutes. The solution was quickly cooled to 20°C, thereafter allowed to drip into stirred deionised water. Sulphonated polymer precipitated as free-flowing beads. Recovery was by filtration, followed by washing with deionised water until the pH was neutral and subsequent drying. By titration the Equivalent Weight was 644.

The sulphonated polymer described above was dissolved in NMP (15%w/w), filtered through 10 micron filter, cast on to a clean glass plate and drawn down, using a Gardner Knife. The solvent was evaporated at 100°C under vacuum for 24 hours producing a membrane of mean thickness of 40 microns. In boiling water the membrane was swollen excessively and broke into pieces.

25

Example 18 - Post treatment of the membrane from Example 17

The procedure of Example 6 was followed, except that the membrane from the Example 17 was immersed in refluxing acetone. In boiling water the membrane remained intact and the % water uptake was 127% as calculated as described above.

Example 19 - Determination of the Crystallinity Index values of Sulphonated Polymers from Examples 2c, 2e and 2g before and after acetone treatment by WAXS

5 Crystallinity can be quantified, in one method, by defining a "crystallinity index" for measurements made by Wide Angle X-ray Scattering (WAXS). This approach defines the measurement in relation to the WAXS pattern. The measured area of crystalline peaks in the WAXS pattern is
10 taken as a percentage of the total crystalline and amorphous scatter over a chosen angular range of the pattern. Crystallinity index should, to a first approximation, be proportional to crystallinity for broadly similar polymer materials. It is constrained to be
15 zero when crystallinity is zero and 100% when crystallinity is 100%.

Membranes of the sulphonated polymers from Examples 2c, 2e and 2g as sulphonated in Example 4, membrane
20 fabricated in Example 5a and post treated with acetone as described in Example 6 were examined by WAXS as described below.

The membranes were analysed using a Siemens D5000 X-
25 ray diffractometer with Cu K-alpha radiation and a Kevex energy dispersive detector. Measurements were made from a single membrane sheet mounted in symmetrical reflection geometry. A programmable divergence slit was used to maintain a constant irradiated region of the specimen
30 surface 6 mm long over a 2-theta measurement range of 10 - 49°.

The WAXS pattern of the membrane from Example 2g before and after acetone treatment exhibited only broad amorphous scatter, whereas the patterns of the membranes from Examples 2c and 2e, before and after acetone treatment exhibited sharper, crystalline peaks in addition to amorphous bands.

The measured WAXS patterns were analysed by first making a background correction, subtracting the corresponding WAXS pattern from a blank specimen holder. The resulting patterns were fitted by a combination of a pattern measured from a similar but amorphous membrane film and a set of peaks (at approximately 18.8, 20.8, 22.9, 29.1 and 40.0 ° 2-theta) corresponding to those observed in the more crystalline membranes. The crystallinity index was calculated as the total area fitted by these peaks taken as a percentage of the combined area of the fitted peaks and the fitted amorphous pattern.

The results are detailed in the Table below.

Sulphonated polymer from Example	Crystallinity Index(%)	
	Before acetone treatment	After acetone treatment
2g	0	0
2e	2	5.5
2c	7	9

Example 20 Determination of the Crystallinity Index values of Sulphonated Polymers from Examples 7c and 7f by WAXS

- 5 Membranes of the sulphonated polymers from Examples 7c and 7f as prepared in Examples 8c and 8f and post treated with acetone in Example 9 were examined by WAXS as described in the previous Example.
- 10 The WAXS pattern of the membrane from Example 7f before and after acetone treatment exhibited only broad amorphous scatter, whereas the patterns of the membrane from Examples 7c before and after acetone treatment exhibited sharper, crystalline peaks in addition to
- 15 amorphous bands.

The results are detailed in the Table below.

Sulphonated polymer from Example	Crystallinity Index(%)	
	Before acetone treatment	After acetone treatment
7f	0	0
7c	6	12

20 Example 21 Comparisons of Methanol Cross-over

- A two compartment, water-jacketed, glass diffusion cell was used to determine the permeability of a membrane to methanol. One compartment of the cell (Compartment A) was charged with an 8 vol% solution of methanol (BDH, Analar), and 0.1 vol% 1-propanol (BDH, Analar) in
- 25 deionised water, the other compartment (Compartment B) was charged with a solution of 0.1 vol% 1-propanol in deionised water. The membrane under investigation was

pretreated by immersion in deionised water for 120 minutes at room temperature then clamped between the two compartments so that the area of membrane in contact with the solutions was 19.63cm^2 . The solutions in both
5 compartments were continually stirred during the experiment by using magnetic followers and the temperature of the cell was maintained at 60°C by means of the water jacket.

10 For each membrane fresh solutions were used in Compartments A and B and the concentration of methanol in Compartment B was measured at regular intervals during the experiment by removing $5\mu\text{l}$ aliquots and analysing the solution by gas chromatography (Varian 3400, GC, fitted
15 with Chrompack 80 - 100 mesh POROPAK Q packed column, using the temperature profile 105°C for 5 minutes, ramping at $10^\circ\text{C}/\text{minute}$ to 130°C and holding for 12 minutes) using 1-propanol as an internal standard. The results are detailed in the Table below

Sulphonated Polymer	Thickness (μm)	EW	Concentration of Methanol in Compartment B (vol%)					
			10 mins	20 mins	30 mins	40 mins	50 mins	60 mins
Nafion 115*	125	1100	0.12	0.33	0.60	0.87	1.16	1.33
Nafion 112	50	1100	0.41	0.72	1.07	1.45	1.81	2.10
Sulphonated polymers before acetone treatment								
Example 2g	40	690	0.09	0.26	0.43	0.61	0.82	0.99
Example 2e	40	670	0.14	0.25	0.41	0.58	0.75	0.90
Example 2c	40	667	0.08	0.21	0.37	0.50	-	0.64
Example 7f	40	583	0.27	0.50	0.89	1.24	1.44	1.83
Example 7c	40	558	0.14	0.34	0.42	0.55	0.85	1.01
Example 7d	40	557	0.13	0.34	0.48	0.64	0.78	0.93
Sulphonated polymers after acetone treatment								
Example 2e	40	670	0.06	0.19	0.29	0.40	0.55	0.60
Example 2c	40	667	0.05	0.14	0.23	0.38	0.50	0.57
Example 7c	40	558	0.13	0.29	0.49	0.67	0.82	0.90
Example 7d	40	557	0.12	0.27	0.41	0.61	0.82	0.91
Example 17	40	644	0.06	0.19	0.29	0.44	0.58	0.70

The reader's attention is directed to all papers and documents which are filed concurrently with or previous to this specification in connection with this application and which are open to public inspection with this specification, and the contents of all such papers and documents are incorporated herein by reference.

All of the features disclosed in this specification (including any accompanying claims, abstract and drawings), and/or all of the steps of any method or

process so disclosed, may be combined in any combination, except combinations where at least some of such features and/or steps are mutually exclusive.

5 Each feature disclosed in this specification (including any accompanying claims, abstract and drawings), may be replaced by alternative features serving the same, equivalent or similar purpose, unless expressly stated otherwise. Thus, unless expressly stated otherwise,
10 each feature disclosed is one example only of a generic series of equivalent or similar features.

The invention is not restricted to the details of the foregoing embodiment(s). The invention extend to any novel
15 one, or any novel combination, of the features disclosed in this specification (including any accompanying claims, abstract and drawings), or to any novel one, or any novel combination, of the steps of any method or process so disclosed.

20

CLAIMS

1. A fuel cell powered by direct fuel (hereinafter a
 "FCDF"), said FCDF having a polymer electrolyte membrane
 5 (PEM) which includes a semi-crystalline polymer.

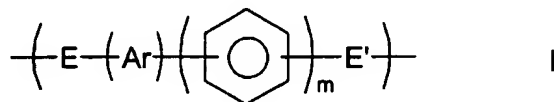
2. A fuel cell according to claim 1, wherein the level of
 crystallinity in said semi-crystalline polymer is at least
 0.5%.

10

3. A fuel cell according to claim 1 or claim 2, wherein
 semi-crystalline polymer includes a repeat unit which
 includes aromatic group containing moieties linked by -CO-
 and/or Q groups, where Q represents -O- or -S-.

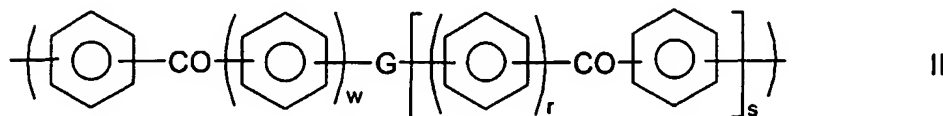
15

4. A fuel cell according to any preceding claim, wherein
 said semi-crystalline polymer is one having a moiety of
 formula

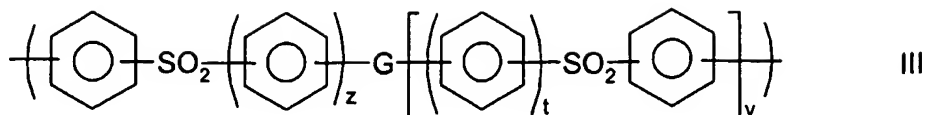


20

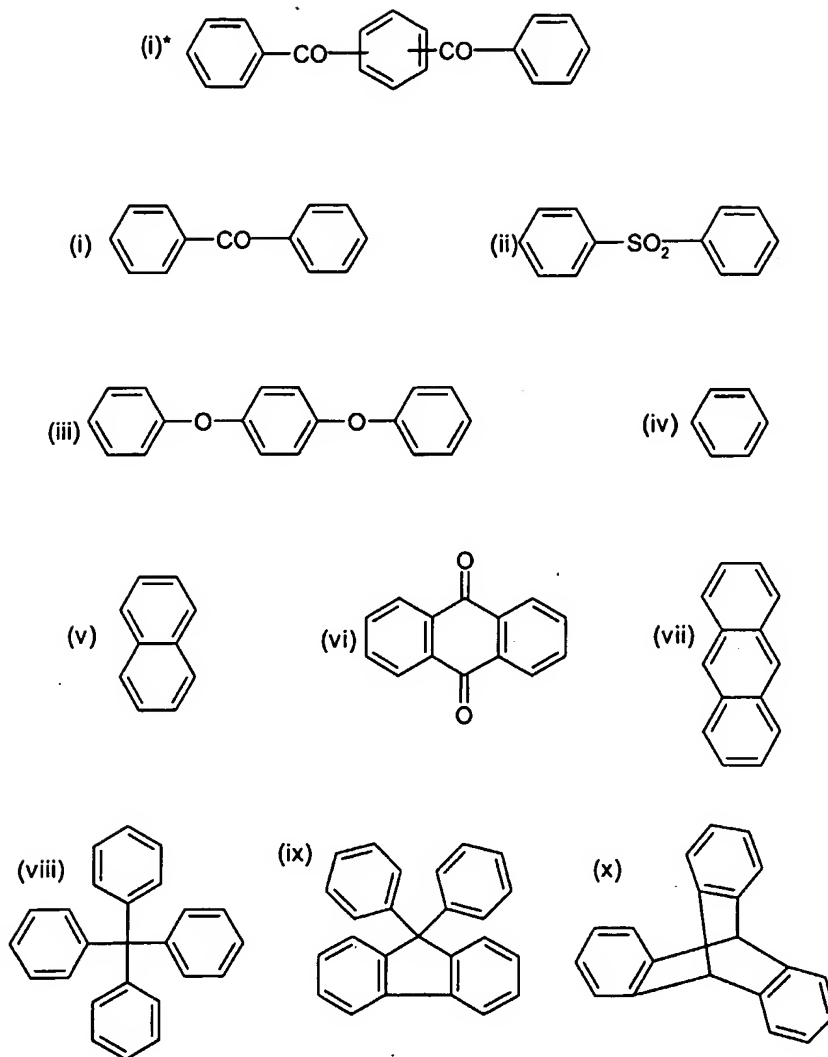
and/or a moiety of formula



25 and/or a moiety of formula



wherein at least some of the units I, II and/or III are
 5 functionalized to provide ion-exchange sites; wherein the
 phenyl moieties in units I, II, and III are independently
 optionally substituted and optionally cross-linked; wherein
 said material is crystallisable and/or semi-crystalline;
 and wherein m,r,s,t,v,w and z independently represent zero
 10 or a positive integer, E and E' independently represent an
 oxygen or a sulphur atom or a direct link, G represents an
 oxygen or sulphur atom, a direct link or a -O-Ph-O- moiety
 where Ph represents a phenyl group and Ar is selected from
 one of the following moieties (i)* or (i) to (x) which is
 15 bonded via one or more of its phenyl moieties to adjacent
 moieties



5 5. A fuel cell according to any preceding claim, wherein
said semi-crystalline polymer is sulphonated.

6. A fuel cell according to any preceding claim, wherein
said semi-crystalline polymer is a copolymer comprising a
10 first repeat unit which is selected from the following:

5 (a) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m and s represent zero, w represents 1 and A and B represent 1;

10 (b) a unit of formula IV wherein E represents an oxygen atom, E' represents a direct link, Ar represents a moiety of structure (i), m represents zero, A represents 1, B represents zero;

15 (c) a unit of formula V wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m and v represent zero, z represents 1 and C and D represent 1;

20 (d) a unit of formula V wherein E represents an oxygen atom, E' represents a direct link, Ar represents a moiety of structure (ii), m represents 0, C represents 1, D represents 0; or

25 (e) a unit of formula V wherein E and E' represents an oxygen atom, Ar represents a structure (i), m represents 0, C represents 1, Z represents 1, G represents a direct link, v represents 0 and D represents 1;

30 (aa) a unit of formula IV wherein E represents an oxygen atom E' represents a direct link, Ar represents a structure (i)*, m represents 0, A represents 1, B represents 0;

(bb) a unit of formula IV wherein E and E' represent oxygen atoms, Ar represents a structure (iv), m and w represent 0, G represents a direct link, s and r represent 1, A and B represent 1;

cc) a unit of formula IV wherein E and E' represent oxygen atoms, Ar represents a structure (i), m and w represent 0, G represents a direct link, s and r represent 1, A and B represent 1;

and a second repeat unit which is selected from the following:

(f) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m represents 1, w represents 1, s represents zero, A and B represent 1;

(g) a unit of formula IV wherein E represents an oxygen atom, E' is a direct link, G represents a direct link, Ar represents a moiety of structure (iv), m and s represent zero, w represent 1, A and B represent 1;

(h) a unit of formula V wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (iv), m represents 1, z represents 1, v represents 0, C and D represent 1; and

(i) a unit of formula V wherein E represents an oxygen atom, E' represents a direct link, G represents a direct link, Ar represents a moiety of structure (iv), m and v represent zero, z represents 1, C and D represent 1;

(j) a unit of formula IV wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (v), m represents 0, w represents 1, s represents 0, A and B represent 1;

(k) a unit of formula V wherein E and E' represent oxygen atoms, G represents a direct link, Ar represents a moiety of structure (v), m represents 0, z represents 1, v represents 0, C and D represent 1.

7. A fuel cell according to any of claims 1 to 5, wherein said semi-crystalline polymer is a copolymer comprising:

a first unit which includes an ion-exchange site;
a second crystalline unit; and, optionally,
a third unit which is amorphous.

25

8. A fuel cell according to claim 7, wherein said first unit is a repeat unit which includes aromatic group containing moieties linked by -SO₂- and/or -CO- and/or -Q- groups, where Q is O or S.

30

9. A fuel cell according to claim 7 or claim 8, wherein said second unit is a repeat unit which includes aromatic

group containing moieties linked by -CO- and/or -Q- groups, where Q is O or S.

10. A fuel cell according to any of claims 7 to 9, wherein
5 said third unit is a repeat unit which includes aromatic group containing moieties linked by -SO₂- and/or -CO- and/or -Q- groups, where Q is O or S provided, however, that said third unit includes a means to render it amorphous (hereinafter said "amorphous means") and/or not
10 crystallisable with polyarylether ketones or polyarylthioether ketones and/or not crystallisable with said second unit.

11. A fuel cell according to any of claims 7 to 10,
15 wherein first units are selected from -ether-phenyl-ketone-phenyl, -ether-phenyl-ketone-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl, -ether-biphenyl-ether-phenyl-ketone-phenyl, ether-phenyl-ether-phenyl-ketone-phenyl, ether-naphthalene-ether-phenyl-ketone-phenyl, ether-phenyl-
20 ether-phenyl-ketone-phenyl-ketone-phenyl, -ether-diphenyl-ether-phenyl-sulphone-phenyl-, -ether-phenyl-ether-phenyl-sulphone-phenyl, and ether-phenyl-sulphone-phenyl functionalised with ion-exchange sites.

25 12. A fuel cell according to any of claims 7 to 11, wherein second units are selected from ether-phenyl-ketone-phenyl-ketone-phenyl-, ether-phenyl-ether-phenyl-ketone-phenyl-ketone-phenyl-, ether-phenyl-ether-phenyl-ketone-phenyl-, ether-phenyl-ketone-phenyl-, ether-phenyl-ketone-phenyl-
30 ether-phenyl-ketone-phenyl-ketone-phenyl and ether-biphenyl-ether-phenyl-ketone-phenyl-.

13. A fuel cell according to any of claims 7 to 12, wherein said third units are selected from ether-phenyl-sulphone-phenyl, ether-phenyl-ether-phenyl-sulphone-phenyl and a - 1,3-dioxy-4-(phenylcarbonyl) phenyl moiety derived from
5 2,4-dihydroxybenzophenone.

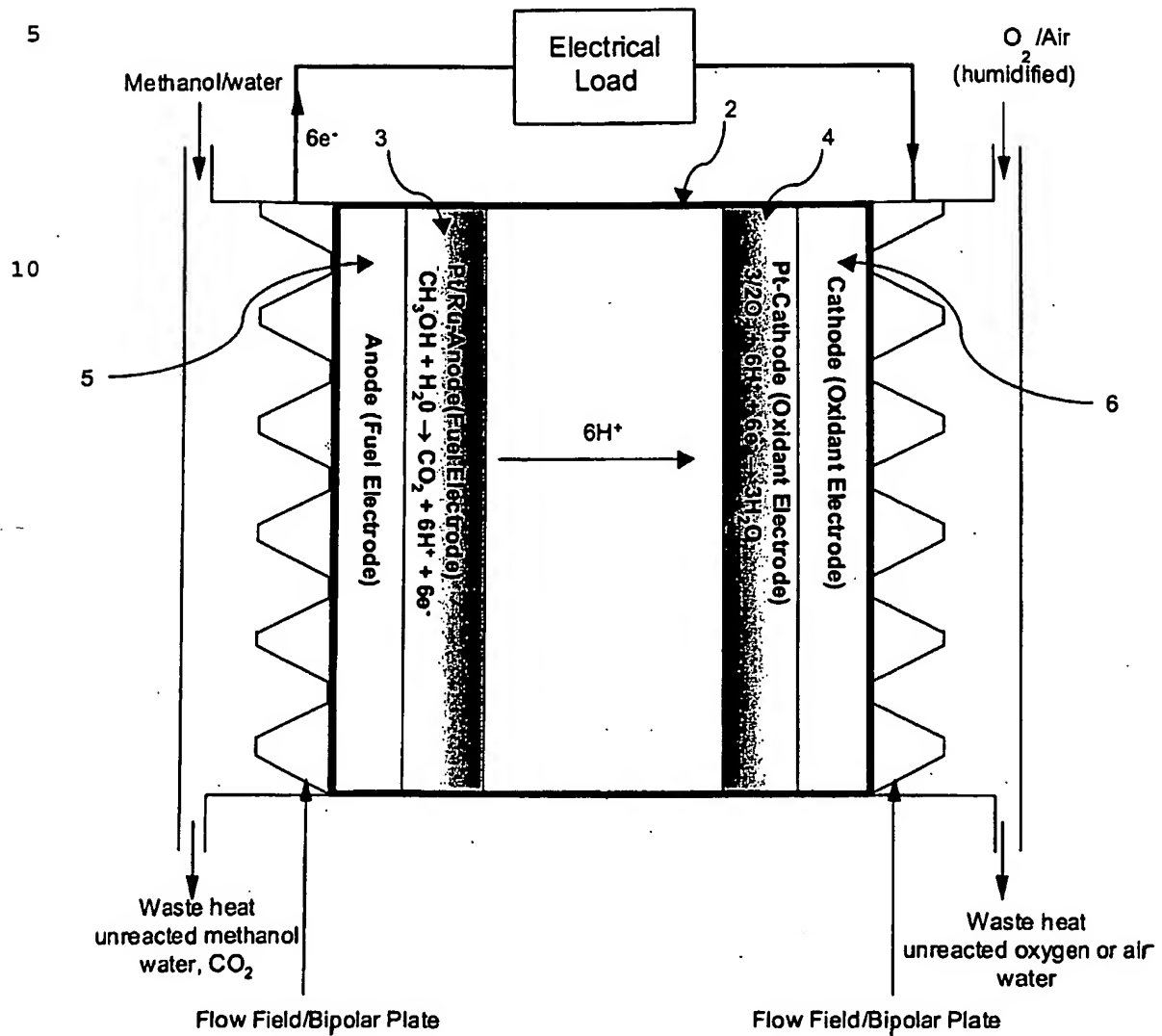
14. A plurality of fuel cells as described in any of claims 1 to 13.

10 15. A method of operating a FCDF, wherein the FCDF is as described in any of claims 1 to 13, the method comprising supplying a fuel to one side of the polymer electrolyte membrane (PEM).

15 16. A method of manufacturing a FCDF, the method comprising incorporating a semi-crystalline polymer into a polymer electrolyte membrane of the fuel cell.

17. The use of a semi-crystalline polymer for the
20 manufacture of a polymer electrolyte membrane of a fuel cell, for reducing the level of fuel cross-over in the cell.

Figure 1



(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization
International Bureau



(43) International Publication Date
26 September 2002 (26.09.2002)

PCT

(10) International Publication Number
WO 02/075835 A3

(51) International Patent Classification⁷: **H01M 8/10**,
B01D 71/06, C08G 65/48, C08J 5/22, H01B 1/12

(21) International Application Number: PCT/GB02/01379

(22) International Filing Date: 21 March 2002 (21.03.2002)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
0107075.4 21 March 2001 (21.03.2001) GB
0123085.3 26 September 2001 (26.09.2001) GB

(71) Applicant (for all designated States except US): **VICTREX MANUFACTURING LIMITED** [GB/GB];
Vitrex Technology Centre, Hillhouse International, P.O.
Box 4, Thornton Cleveleys, Lancashire FY5 4QD (GB).

(72) Inventors; and

(75) Inventors/Applicants (for US only): **ANDREWS, Mark, James** [GB/GB]; 9 Thorntrees Avenue, Lea, Preston PR2 1PJ (GB). **LOCKLEY, John, Edward** [GB/GB]; 10 Rossmoyne Road, Lancaster LA1 4SN (GB). **WILSON, Brian** [GB/GB]; 1 White Lea, Cabus, Garstang, Lancashire PR3 1JG (GB).

(74) Agents: **NEILL, Alastair, William et al.**; Appleyard Lees, 15 Clare Road, Halifax HX1 2HY (GB).

(81) Designated States (*national*): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.

(84) Designated States (*regional*): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Published:

- with international search report
- before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

(88) Date of publication of the international search report:
16 October 2003

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



WO 02/075835 A3

(54) Title: POLYMER ELECTROLYTE MEMBRANE FUEL CELL

(57) Abstract: A fuel cell powered by direct fuel, for example a direct methanol fuel cell, includes a polymer electrolyte membrane which includes a semi-crystalline polymer. Preferred semi-crystalline polymers include first repeat units comprising sulphonated aromatic group containing moieties linked by -SO₂- and/or -CO- and/or -Q- groups, where Q is O or S and second repeat units which include aromatic group containing moieties linked by -CO- and/or Q groups.

INTERNATIONAL SEARCH REPORT

Internat. application No

PCT/GB 02 01379

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 H01M8/10 B01D71/06 C08J5/22 C08G65/48 H01B1/12

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 H01M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 00 15691 A (CHARNOCK PETER ;VICTREX MFG LTD (GB); WILSON BRIAN (GB); KEMMISH D) 23 March 2000 (2000-03-23) cited in the application * see p.20, l. 11-19, p.28, l.1-7, p.30, l.13-16, claims 28 and 35 * the whole document	1-17
X	WO 98 22989 A (CALIFORNIA INST OF TECHN ;UNIV SOUTHERN CALIFORNIA (US)) 28 May 1998 (1998-05-28) * see p.13, l. 3-17,p.34, l.13-p35, l.5, claims 1-6 * the whole document	1-17

-/--

☒ Further documents are listed in the continuation of box C.☒ Patent family members are listed in annex.

* Special categories of cited documents:

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *G* document member of the same patent family

Date of the actual completion of the international search

22 July 2003

Date of mailing of the international search report

06/08/2003

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Stellmach, J

INTERNATIONAL SEARCH REPORT

Internat pplication No

PCT/GB 02 01379

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WO 96 29752 A (GROT WALTHER GUSTAV ;RAJENDRAN GOVINDARAJULU (US); DU PONT (US)) 26 September 1996 (1996-09-26) * see p.2, l.16-26, claims 8-10, 18, and 19 * the whole document ---	1-17
Y	WO 97 19480 A (GRAHAM EVA ;HALPERT GERALD (US); YAVROUIAN ANDRE (US); CALIFORNIA) 29 May 1997 (1997-05-29) cited in the application * see claims 17-22 * the whole document ---	1-17
Y	WO 98 50457 A (SPETHMANN JEFFREY E ;KEATING JAMES T (US)) 12 November 1998 (1998-11-12) * see claims * the whole document ---	1-17
Y	EP 0 688 824 A (HOECHST AG) 27 December 1995 (1995-12-27) * see claims * the whole document ---	1-17
Y	DE 198 47 782 A (AVENTIS RES & TECH GMBH & CO) 20 April 2000 (2000-04-20) * see p.2, l. 6-28, claims * the whole document ---	1-17
Y	KUVER A ET AL: "Comparative study of methanol crossover across electropolymerized and commercial proton exchange membrane electrolytes for the acid direct methanol fuel cell" ELECTROCHIMICA ACTA, ELSEVIER SCIENCE PUBLISHERS, BARKING, GB, vol. 43, no. 16-17, 29 May 1998 (1998-05-29), pages 2527-2535, XP004134162 ISSN: 0013-4686 * see p.2528, left col., 2nd par. ;p.2534, left col., first par. * the whole document ---	1-17
Y	HEINZEL,A. ET AL.: "A review of the state-of-the-art of the methanol crossover in direct methanol fuel cells " JOURNAL OF POWER SOURCES, vol. 84, 1999, pages 70-74, XP002248491 AMSTERDAM * see p.72-74, par. 4 * the whole document ---	1-17

-/--

INTERNATIONAL SEARCH REPORT

Internati Application No
PCT/GB 02 01379

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WASMUS, S. ET AL.: "Methanol oxidation and direct methanol fuel cells: a selective review" JOURNAL OF ELECTROANALYTICAL CHEMISTRY, vol. 461, 1999, pages 14-31, XP002248492 AMSTERDAM * see p.25-26, par. 3.4 electrolytes * the whole document ---	1-17
P, X	WO 01 70857 A (BRIDGES RICHARD FRANK ;CHARNOCK PETER (GB); VICTREX MFG LTD (GB);) 27 September 2001 (2001-09-27) * see claims * the whole document ---	1-17
P, X	WO 01 71839 A (BRIDGES RICHARD FRANK ;LOCKLEY JOHN EDWARD (GB); CHARNOK PETER (G) 27 September 2001 (2001-09-27) * see claims * the whole document -----	1-17

INTERNATIONAL SEARCH REPORT

International application No.

PCT/GB 02 01379

Box I Observations where certain claims were found unsearchable (Continuation of Item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:
because they relate to subject matter not required to be searched by this Authority, namely:

2. ☒ Claims Nos.:
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:

see FURTHER INFORMATION sheet PCT/ISA/210

3. ☐ Claims Nos.:
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

Box II Observations where unity of invention is lacking (Continuation of Item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.

2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.

3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:

4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☐ No protest accompanied the payment of additional search fees.

FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box I.2

Present claims 1-17 relate to an extremely large number of possible semi-crystalline polymers in polymer electrolyte membranes. Support within the meaning of Article 6 PCT and/or disclosure within the meaning of Article 5 PCT is to be found, however, for only a very small proportion of the products claimed. In the present case, the claims so lack support, and the application so lacks disclosure, that a meaningful search over the whole of the claimed scope is impossible. Consequently, the search has been carried out for those parts of the claims which appear to be supported and disclosed, namely those parts relating to the polyarylether ketones and polyarylether sulphones described in the examples and closely related compounds in polymer electrolyte membrane materials and methods.

The applicant's attention is drawn to the fact that claims, or parts of claims, relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure.

INTERNATIONAL SEARCH REPORT

ation on patent family members

Internal Application No

PCT/GB 02 01379

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 0015691	A	23-03-2000	AU 5750999 A CA 2343184 A1 EP 1112301 A1 WO 0015691 A1 JP 2002524631 T	03-04-2000 23-03-2000 04-07-2001 23-03-2000 06-08-2002
WO 9822989	A	28-05-1998	US 6444343 B1 AU 729900 B2 AU 7303998 A EP 0948827 A1 JP 2001504636 T WO 9822989 A1	03-09-2002 15-02-2001 10-06-1998 13-10-1999 03-04-2001 28-05-1998
WO 9629752	A	26-09-1996	DE 69608793 D1 DE 69608793 T2 EP 0815606 A1 JP 2000516014 T WO 9629752 A1 US 5919583 A	13-07-2000 01-02-2001 07-01-1998 28-11-2000 26-09-1996 06-07-1999
WO 9719480	A	29-05-1997	US 5795496 A AU 1406897 A CA 2238189 A1 EP 0870340 A1 JP 2000501223 T WO 9719480 A1	18-08-1998 11-06-1997 29-05-1997 14-10-1998 02-02-2000 29-05-1997
WO 9850457	A	12-11-1998	AU 7271698 A US 6110333 A WO 9850457 A1	27-11-1998 29-08-2000 12-11-1998
EP 0688824	A	27-12-1995	DE 4422158 A1 EP 0688824 A2 JP 8020716 A US 5834566 A	04-01-1996 27-12-1995 23-01-1996 10-11-1998
DE 19847782	A	20-04-2000	DE 19847782 A1 BR 9914597 A CA 2347090 A1 CN 1328484 T WO 0023177 A1 EP 1124625 A1 JP 2002527233 T	20-04-2000 26-06-2001 27-04-2000 26-12-2001 27-04-2000 22-08-2001 27-08-2002
WO 0170857	A	27-09-2001	AU 3940501 A CA 2402838 A1 EP 1265951 A2 WO 0170857 A2	03-10-2001 27-09-2001 18-12-2002 27-09-2001
WO 0171839	A	27-09-2001	AU 4430301 A CA 2402841 A1 EP 1275164 A2 WO 0171839 A2	03-10-2001 27-09-2001 15-01-2003 27-09-2001